

Chapter 4: Carbon

AP Biology 2011

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Backbone of Molecules

- ✦ All living organisms are made up of chemicals based mostly on carbon
- ✦ Compounds based on carbon are called organic compounds
 - ✦ Can range from simple molecules like methane (CH_4) to complex proteins with masses in excess of 100,000 daltons
- ✦ Organic chemistry is the study of compounds that contain carbon

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Synthesis of Organic Compounds

- ✦ Stanley Miller demonstrated abiotic synthesis of organic compounds
- ✦ Once thought that organic compounds could only be made by living things
- ✦ Proven now that under the right conditions, organic compounds can synthesize spontaneously

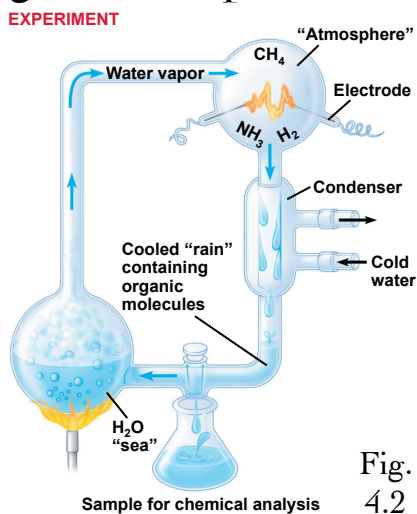


Fig.
4.2

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Bonding

- ✦ Electron configuration is key to an atom's characteristics (determines the kinds and number of bonds an atom will form with other atoms)
- ✦ Carbon has 2 electrons in its first shell and 4 in its outer shell
- ✦ Because of this it can form 4 bonds (tetravalence)
- ✦ Bond angles 109.5°

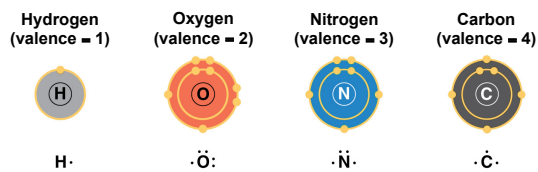
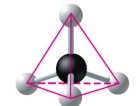

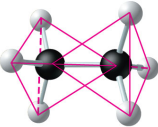

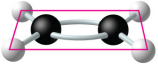



Fig. 4.4

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Carbon Bonding Versatility

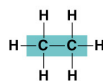
Name and Comment	Molecular Formula	Structural Formula	Ball-and-Stick Model	Space-Filling Model
(a) Methane	CH_4	<pre> H H-C-H H </pre>		
(b) Ethane	C_2H_6	<pre> H H H-C-C-H H H </pre>		
(c) Ethene (ethylene)	C_2H_4	<pre> H H \ / C=C / \ H H </pre>		

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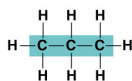
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Variations in Carbon Chains

(a) Length

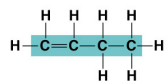


Ethane

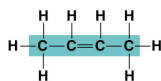


Propane

(c) Double bond position

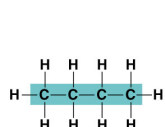


1-Butene

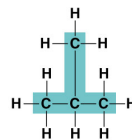


2-Butene

(b) Branching

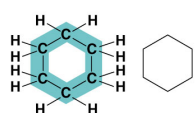


Butane

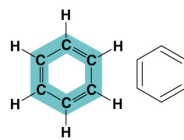


2-Methylpropane
(isobutane)

(d) Presence of rings



Cyclohexane



Benzene

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Fig. 4.5

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Variations in Carbon Chains

- Hydrocarbons - organic molecules consisting of only carbon and hydrogen
 - Major component of petroleum
 - Major components of fats

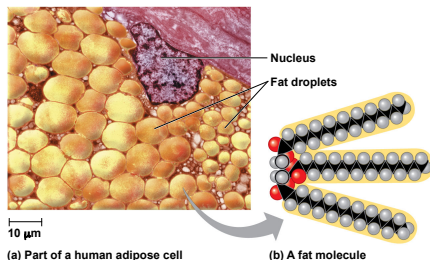
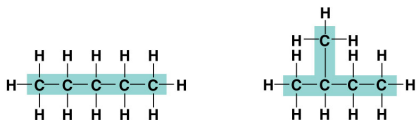


Fig. 4.6

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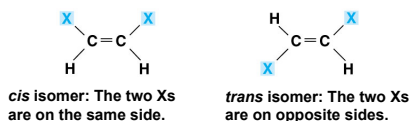
(a) Structural isomers



Isomers

- Isomers - compounds that have the same numbers of atoms but different structures and different properties

(b) Cis-trans isomers



(c) Enantiomers

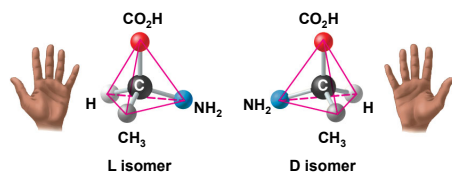


Fig. 4.7

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Enantiomers

Drug	Condition	Effective Enantiomer	Ineffective Enantiomer
Ibuprofen	Pain; inflammation	 S-Ibuprofen	 R-Ibuprofen
Albuterol	Asthma	 R-Albuterol	 S-Albuterol

Fig. 4.8

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Functional Groups

- Functional groups are most important in chemistry of life
- Each functional group behaves the same regardless of which molecule it is on
- Give each molecule unique properties

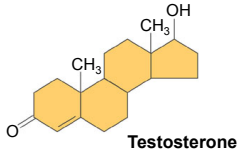
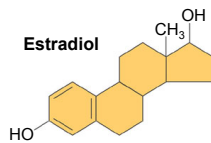


Fig. 4.9 Female lion
Male lion

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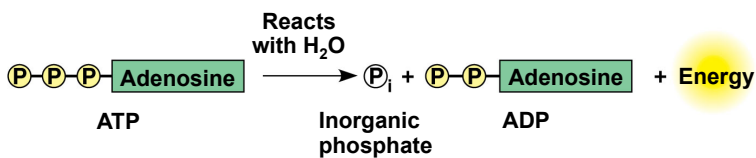
Six Key Functional Groups

- Hydroxyl
- Carbonyl
- Carboxyl
- Amino
- Sulfhydryl
- Phosphate
- Methyl

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ATP

- Source of energy in cellular processes
- Adenosine triphosphate
- Consists of:
 - Adenosine (organic molecule)
 - Three phosphates
- Energy released by stripping a phosphate to become ADP



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