

# CHAPTER 18: AIR POLLUTION

APES 2013

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# THE ATMOSPHERE TROPOSPHERE - CLOSEST TO THE EARTH'S SURFACE, MAINLY COMPOSED OF NITROGEN (78%) AND OXYGEN (21%) STRATOSPHERE - SIMILAR TO THE TROPOSPHERE WITH LESS WATER VAPOR AND MORE OZ.ONE OZ.ONE LAYER - FOUND IN STRATOSPHERE AND ACTS AS A UV LIGHT FILTER (302 + UV = 203)







SECONDARY POLLUTANTS - PRIMARY POLLUTANTS REACT WITH ONE ANOTHER AND WITH COMPONENTS OF THE AIR TO FORM NEW CHEMICALS

# MAJOR POLLUTANTS: CARBON OXIDES

Pollutant	Characterístics	Sources	Effects
Carbon Monoxide (CO)	colorless, odorless, forms during incomplete combustion of carbon- containing compounds	vehicle exhaust, burning of forests and grasslands, tobacco smoke	reacts w/ hemoglobin in red blood cells, trigger heart attacks, headaches, nausea drowsiness, menta impairments
Carbon Dioxide (CO <sub>2</sub> )	colorless, odorless, forms during combustion of carbon- containing compounds	any combustion of organic material	climate change

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# MAJOR POLLUTANTS: NITROGEN BASED

Pollutant	Characterístics	Sources	Effects
Nitrogen Oxides (NO, NO×)	colorless, forms when nitrogen and oxygen react at high temps (ex. automobile engine, coal power plant, lightning)	vehicle exhaust, coal power plants, lighting strikes, nitrogen fixing bacteria	irritates eyes, nose, and throat; respiratory aggravation, suppress plant growth
Nitric Acid (HNO3)	COMPONENTS: Nitrogen oxides (see above) and water	nitrogen oxides react with water vapor	acid deposition; formation of smog; irritates eyes, nose, and throat; respiratory aggravation, suppres: plant growth

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Pollutant	Characterístics	Sources	Effects
Sulfur Dioxide (SO <sub>2</sub> )	colorless gas with strong oder	One third from natural sources (sulfur cycle) Majority from coal power plants, oil refining, smelting ores	Asian Brown Cloud; damage plants, soil, and aquatic life; damage stone, paint, metal
Sulfuric Acid (H2SO4)	corrosive	sulfur dioxide reacting with water vapor	damage plants, soil, and aquatic life; damage stone, paint, metal; aggravate breathing problems

# MAJOR POLLUTANTS: PARTICULATES

Pollutant	Characterístics	Sources	Effects
Suspended Particulate Matter (SPM)	variety of solid particles and liquid droplets Types: Fine particles (diameter less than 10 micrometers), Ultrafine Particles (diameter less than 2.5 micrometers - less than one quarter a human hair)	Natural sources: dust, wild fires, sea salt Human sources: coal power plants, vehicles, smoke	In general: irritate nose and throat, damage lungs, aggravate asthma Toxics (lead, cadmium, and PCBs): cause mutations, reproductive problems, cancer

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Pollutant	Characterístics	Sources	Effects
Ozone (O3)	colorless, highly- reactive gas, main component of smog	reaction between sunlight and air containing hydrocarbons and nitrogen oxides	coughing and breathing problems; aggravat heart and lung disease; irritate eyes, nose, and throat; damages plants, rubber, fabric, and paint

MAJOR POLLUTANTS: VOCs Pollutant Characterístics Sources Effects Short term Natural sources: exposure: dizziness, plants, wetlands organic compounds unconsciousness, that exist as gasses in Human sources: rice Volatile death the atmosphere: paddies, landfills, oil Organic isoprenes and terpenes and natural gas wells, Long term Compounds (emitted by plants), cows, tobacco exposure: methane leukemia, blood smoke, plastics, rubber, dry cleaning disorders, immune fluid, paint system damage







# NATURAL FACTORS THAT REDUCE AIR POLLUTION

- HEAVY PARTICLES SETTLE OUT BECAUSE OF GRAVITY
- RAIN AND SNOW PULL PARTICLES OUT OF THE AIR
- SALTY SPRAY WASHES OUT PARTICLES OUT OF AIR BY SEA
- WIND BLOWS AND MIXES POLLUTED AND CLEAN AIR
- CHEMICAL REACTIONS CAN REMOVE

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# FACTORS THAT INCREASE AIR POLLUTION URBAN BUILDINGS SLOW WIND SPEED AND REDUCE DILUTION HILLS AND MOUNTAINS REDUCE THE FLOW OF AIR IN VALLEYS BELOW THEM HIGH TEMPERATURES PROMOTE REACTIONS THAT FORM

- HIGH TEMPERATURES PROMOTE REACTIONS THAT FORM SMOG
- VOLATILE ORGANIC COMPOUNDS
- GRASSHOPPER EFFECT POLLUTANTS TRANSPORTED BY EVAPORATION AND WINDS FROM TROPICAL AND TEMPERATE AREAS TO THE POLAR AREAS
- GROUND-LEVEL AIR BECAUSE AIR DOES NOT MIX



# ACID DEPOSITION

### SMOKESTACKS

- $\Box\,\,$  Emit sulfur dioxide, SPM, and nitrogen oxides high into the ATMOSPHERE SO THE WIND CAN DILUTE AND DISPERSE
- REDUCE LOCAL AIR POLLUTION; INCREASE REGIONAL AIR POLLUTION
- □ SUBSTANCES CAN STAY IN THE AIR FOR 2-14 DAYS
- ACID DEPOSITION (POLLUTION FALLS TO THE GROUND AS)
  - $\Box$  wet deposition Acidic RAIN, snow, or Fog (PH < 5.6)
  - DRY DEPOSITION ACIDIC PARTICLES









 REDUCING ACID DEPOSITION

 PREVENTION APPROACHES

 LOW SULFUR COAL

 REMOVING SULFUR FROM COAL BEFORE BURNING

 ALTERNATIVE ENERGY SOURCES

 TREATMENT

 LIMESTONE USED TO NEUTRALIZE ACIDIFIED LAKES AND SOIL

# INDOOR AIR POLLUTION BIG PROBLEM IN DEVELOPED COUNTRIES POLLUTANT LEVELS CAN BE 2 TO 5 TIMES HIGHER THAN OUTSIDE POLLUTANT LEVELS IN CARS IN URBAN AREAS CAN BE 18 TIMES HIGHER HEALTH RISKS ARE HIGHER BECAUSE 70-98% OF TIME INDOORS







# BODY DEFENSES

- MUCUS AND CILIA IN RESPIRATORY TRACT
- □ EMPHYSEMA CAN RESULT FROM CONTINUED EXPOSURE TO POLLUTANTS AND SMOKING



FIG. 18-20



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# LAWS AND REGULATIONS CLEAN AIR ACTS IN 1970, 1977, AND 1990 FPA DEVELOPED NAAQS (NATIONAL AMBIENT AIR QUALITY STANDARDS) FOR SIX POLLUTANTS: CARBON MONOXIDE, NITROGEN OXIDES, SULFUR DIOXIDES, SPM, OZONE, AND LEAD EPA EMISSION STANDARDS FOR 188 HAPS (HAZARDOUS AIR POLLUTANTS) TRI (TOXIC RELEASE INVENTORY) - REQUIRES INDUSTRIES TO REPORT THEIR RELEASES AND WASTE MANAGEMENT FOR 667 DIFFERENT CHEMICALS THIS WAS WEAKENED IN 2005

# WHY US SUCCESSES

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# WEAKNESSES OF AIR POLLUTION LAWS

- LAWS RELY ON CLEAN-UP MORE THAN PREVENTION
- HAVE NOT INCREASED FUEL EFFICIENCY STANDARDS
- □ INADEQUATE EMISSIONS CONTROLS FOR MOTORCYCLES AND TWO-CYCLE ENGINES
- LACKS REGULATION FOR SHIPS IN AMERICAN PORTS
- AIRPORTS ARE EXEMPT
- $\Box$  ~ No regulation of greenhouse gas  $co_2$
- U ULTRAFINE PARTICLES NOT REGULATED
- URBAN OZONE LEVELS STILL TOO HIGH
- LAWS DO NOT DEAL WITH INDOOR AIR POLLUTION
- □ REGULATIONS OF CLEAN AIR ACT NOT ALWAYS ENFORCED

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# USING CAPITALISM TO REDUCE AIR POLLUTION

- CLEAN AIR ACT OF 1990 ANTHORIZES EMISSIONS TRADING OR CAP-AND-TRADE PROGRAMS
  - ALLOWS POLLUTERS TO BUY AND SELL POLLUTION RIGHTS
  - EACH POLLUTER GETS CREDITS AND CAN SELL THEM TO ANOTHER PLANT
  - □ THIS WILL HELP REDUCE POLLUTION IF THE AMOUNT OF CREDITS GIVEN OUT CONTINUES TO DECREASE