

Focus Questions and Key Terms
Chapters 2-3
Biochemistry
Honors Biology

Focus Questions

Chapter 2

- List the four most common elements. (2.1)
- Explain the importance of the trace elements iron and iodine. (2.2)
- Describe the properties of sodium, chloride, and sodium chloride. (2.3)
- Explain what isotopes are and the different ways they can be helpful and harmful. (2.4-2.5)
- Describe the difference between ionic, covalent, and hydrogen bonds. Provide examples of each. (2.6-2.10)
- Explain the special properties hydrogen bonds give water. (2.11-2.15)
- Explain how acid precipitation impacts ecosystems. (2.16)

Chapter 3

- What is lactose and how does the human body process it? (pg. 33 and 3.17)
- Why are functional groups important? Draw the six discussed in the book. (3.2)
- Draw a simple dehydration reaction. What is this process used to do? (3.3)
- Draw a simple hydrolysis reaction. What is this process used to do? (3.3)
- Explain why high-fructose corn syrup is thought to be responsible for higher rates of obesity. (3.6)
- Explain how the following polysaccharides are used: starch, glycogen, cellulose, and chitin. (3.7)
- Explain how fats are used in biological systems. (3.8-3.9)
- Explain how proteins are used in biological systems. (3.11)
- Explain how a protein's shape determines its function. (3.13-3.14)
- Explain how nucleic acids are used in biological systems. (3.16)

Key Terms

Chapter 2

acid	electron	polar covalent bond
acid precipitation	electronegativity	polar molecule
adhesion	element	product
atom	hydrogen bond	proton
atomic mass	ion	radioactive isotope
atomic number	ionic bond	reactant
base	isotope	solute
buffer	matter	solution
chemical bond	molecule	solvent
chemical reaction	neutron	surface tension
cohesion	nonpolar covalent bond	temperature
compound	nucleus	trace element
covalent bond	pH scale	

Chapter 3

amino acid	gene	polymer
anabolic steroid	glycogen	polypeptide
carbohydrate	hydrolysis	polysaccharide
carbon skeleton	hydrophilic	protein
cellulose	hydrophobic	ribonucleic acid
chitin	isomers	saturated
cholesterol	lipid	starch
dehydration reaction	macromolecule	steroid
denaturation	monomer	unsaturated
deoxyribonucleic acid	monosaccharide	
disaccharide	nucleic acid	
double helix	nucleotide	
enzyme	organic compound	
fat	peptide bond	
functional group	phospholipid	